

A123 – Petroleum-Testing Guidance/Interpretation
Revision 1.1 – February 2009



CALA

Laboratory Accreditation

Introduction

This guidance/interpretation document is for use in conjunction with A120 – *Petroleum-Testing Checklist*, when assessing methods specifically designed for testing in the petroleum industry. Where there are differences between the checklist, this guidance/interpretation document, the reference method, and ISO/IEC 17025, the requirements in the reference method and ISO/IEC 17025 shall prevail.

The item numbers noted below correspond to the item numbers in A120 – *Petroleum-Testing Checklist*.

Guidance/Interpretation

01.01 (Authorized test method)

Some common petroleum-testing references include, but are not limited to, ASTM, GPA, and UOP.

02.01 (Method validation)

ASTM methods normally state expected method precision on two bases: “Reproducibility” and “repeatability”. Reproducibility is the expected inter-laboratory variance.

ASTM methods include precision statements to provide a guideline on the variability expected between results obtained under prescribed conditions and assess, in general terms, its validity for specific applications. The repeatability (within lab variability) should normally be the *basis* for the lab repeatability in method validation/verification. However, these performance levels may not always be achievable unless using the same materials (e.g. performance levels determined from refined products may not be achievable for crude oils. It is not necessary that the lab achieve the repeatability performance specified in the reference, but it should use this as a guideline for expected variability.

Reproducibility (between lab variability) is probably useful in test method selection and service validation.

02.01 (Measurement Uncertainty)

P19 (CALA Policy on the Estimation of Uncertainty of Measurement in Environmental Testing) covers estimation of MU for most petroleum-testing

methods. For methods used in petroleum-testing that are more or less empirical, PT variability provide a reliable, comprehensive, quick and easy estimate.

Testing for compliance to a specification is common for many petroleum-testing methods. Gases and liquids transported in pipelines must meet certain specifications (this may be done to protect the pipeline or other vessels from corrosion etc.) There may be any number of specifications around custody transfer points. These are usually contractual and can vary from simple heating value / moisture content to more exotic concerns such as specific contaminants such as PCBs, metals, benzene. Per the standard, this has implications related to requirements for **measurement uncertainty** (client knowledge related to uncertainty is, as with most industries, limited). The lab can, as service, have policy and procedure related to notifying clients of nonconformance to product specification and using specifications for data verification / validation.

03.01 (Test method details)

Sample pretreatment is important for many petroleum methods (e.g. cleaning oil, single phasing HP liquids). Most tests on crude oil are performed on oil cleaned by centrifugation and are reported on an “after cleaning” basis.

Pressurized hydrocarbon samples may be anything from light gas mixtures consisting mainly of methane to gasified crude oils with everything from C1 to C30+ hydrocarbons. The latter are problematic to analyze since the components represent a very wide range of boiling points. Labs may have various approaches including separating the sample into gas and liquid phases stable at ambient conditions, measuring each phase for composition and volume or mass, then recombining these mathematically back into the original mixture.

Identification and measurement of individual hydrocarbon components is usually impractical in chromatographic analysis of hydrocarbon mixtures. Analysis usually involves grouping of peaks by retention time that approximates boiling point, which approximates carbon number.

Measurement is usually made on a mass basis but reported on mass, volume and mole bases. Conversion requires molecular weight and density that, in principle, can be estimated from the chromatographic analysis. However, because of the approximations indicated in the previous paragraph, and the error in assigning properties to a group based on carbon / boiling point these estimates can seriously misrepresent the “true” result. These parameters can be measured, but normally only on the stabilized phase. It is important that the lab use a valid approach to calculation and that the calculations be

compatible with the bases of all measurements made.

Data reduction is critical in some cases. The most important results for valuation of natural gas (density and heating value) are not measured directly, but are calculated from composition. If the scope is to include these, then the data reduction must be detailed.

Quite a bit of data validation normally goes on with pressurized samples for hydrocarbon composition (e.g. comparison to pipeline specs, equilibrium k-plots, calculated vs. measured properties such as density of gas, comparison to historical data, methane vs. pressure on liquids).

Normally for production testing:

Equilibrium k-plots (or equivalent) should be used i) to verify phase separation (if this is done) and ii) to verify sample integrity from associated gases and liquids (e.g. from an inlet separator). K-plots are thermodynamic diagrams based on the Clausius-Clapeyron equation indicate whether the gas and liquid are at equilibrium and can verify if results are mutually consistent.

It is recommended that historical data should be used where available and where applicable.

The lab should have protocols for comparing related parameters for compatibility / validity / expected behaviour. These can be such things as:

- A reasonably smooth variation of concentration vs. carbon number in natural materials.
- Historical consistency (where applicable)
- Geological and geographical consistency (fluids from the same geological pool in the same area can be expected to be broadly consistent).
- Measured properties vs. the same property calculated from composition (e.g., gas density).

04.02 (Sampling and sub-sampling)

The sampling process is critical for pressurized samples. There are reference methods for gas sampling (GPA has this, so does ISO). They are generally applicable but may not be applicable in all situations depending on the source composition, source conditions, ambient conditions etc. Sampling procedures that might work for lean gases may bias rich gases.

If accreditation is going to cover sample collection, the lab should be following good practices and have validated their sampling protocols for the types of samples collected. Samplers would have to be appropriately trained, be able to recognize potential problems, etc.

Hydrocarbons are generally sampled based on availability at agreed upon points using techniques based on sound physical principles and good practice. Usually samples will be collected from a single specific point along a flowing stream, variation is normally with time. “Statistical” sampling plans would therefore require a model of variation in the source over *time* and *flow*. “Spot” samples are collected at a single point in time and may be taken to represent the source over an assumed interval (this is quite reasonable if the stream is consistent). “Proportional” samples use special sampling equipment to collect sample in proportion with time or with flow rate. This technique more costly, labour intensive and complex, but is more representative than spot samples or statistical techniques. Either way, this is normally based on agreement between lab and client; clients normally manage their own proportional sampling.

Generating representative samples is not a problem if dealing with a stable fluid. With high-pressure liquids, phase changes can occur which can result in bias if not controlled.

The validity of sampling and sub-sampling methods can be critical if dealing with multiphase samples. Sample integrity must especially be evaluated when the samples are gas and high-pressure liquids because opening pressures, air contents, bubble points etc. establish whether the sample can be representative of the source.

The lab should be checking opening pressures / temperature of gas and liquid samples and should have nonconformance criteria and actions to taken for nonconformance based on source temperature and pressure.

The lab should measure O₂ content of gas samples. O₂ does not occur in natural gas and is normally assumed an artifact of sampling (due to air contamination). The lab should have a protocol for dealing with O₂. Normally, small quantities of air can be corrected out, O₂ and an amount of N₂ in proportion to that found in air, is taken from the result and the remainder renormalized - air free basis. Larger amounts of air indicate a seriously compromised sample and should normally rejected or - at minimum - a note added on the report.

Care must be taken when measuring and correcting for O₂ as Ar may co-elute. There are ways of dealing with this (Ar carrier, cryo temperatures, suitable

accounting in calculation...) the lab should have the means clearly defined.

Bubble points (essentially measuring the conditions required to return a sample to single - liquid - phase) are useful as well in determining sample integrity.

Many sample containers used in the oil and gas sector are NOT disposable and get cleaned and reused. Therefore, the lab must have a procedure for this step.

Auto samplers in use must meet requirements of ASTM D4177, or the frequency and volume of manual samples is varied to meet ASTM 4057 requirements to avoid product representation issues.

05.01 (Calibration)

In many situations, calibration is more that of a piece of measurement equipment than the method. This may be a basic physical measurement such as temperature or pressure, or equipment such as viscometers, which may require external/manufacturer calibration.

Calibration for hydrocarbon composition by GC may involve determining relative response factors (RRF). A multilevel calibration may be redundant in situations where RRF's are used. RRF's may even be stipulated in reference methods.

09.01 (Equipment)

Of relevance beyond device calibration is the device itself. Some of these methods are empirical. Most (e.g. ASTM) are quite detailed in measurement equipment requirements and configuration. If there are any modifications to equipment, these need to be both documented (clearly stated in the lab SOP) and verified that the changes are within the allowance specified in the referenced method.

10.01 (Supplies)

The containers themselves should be stainless steel and capable of withstanding the sample pressure. They should also be suitable to the sampling technique (certain liquid displacement sampling methods require special equipment configuration). Carbon steel should not be used, it absorbs CO₂ and it corrodes so the integrity of the vessel can fail. Even stainless steel is inappropriate for sampling gases for trace sulfur species as these may be absorbed. Tedlar sample bags are typically used for reduced sulfur species.

On the other hand, samples in Tedlar are generally not suitable for hydrocarbon composition analysis due to permeability to some constituents, and other effects that may result in bias.

11.01 (Record-keeping)

When measuring gases, it is important that conditions (temperature & pressure) be appropriately measured and compensated.

Volume correction to reference conditions can be critical, especially for gases. If necessary, the lab should have an appropriate means of measuring Boiling Point.

With gases, it is critical to report the reference conditions and the basis of the measurement. There are at least a dozen different sets of “standard conditions”. They should be stated directly and unambiguously (The *Electricity and Gas Inspection Act* of Canada specify 101.325 kPa and 15 °C which is slightly different from the conditions used by some US bodies). Use of “STP” or “NTP” is not acceptable.

Although gases are normally measured “as received”, properties can be reported on different bases such as “moisture free” or “moisture and acid gas free”. Both are valid, but these bases must be clearly indicated.

There are a few tests that may only be valid if done directly at the source (e.g. dew points by the chilled mirror method ASTM D1142).